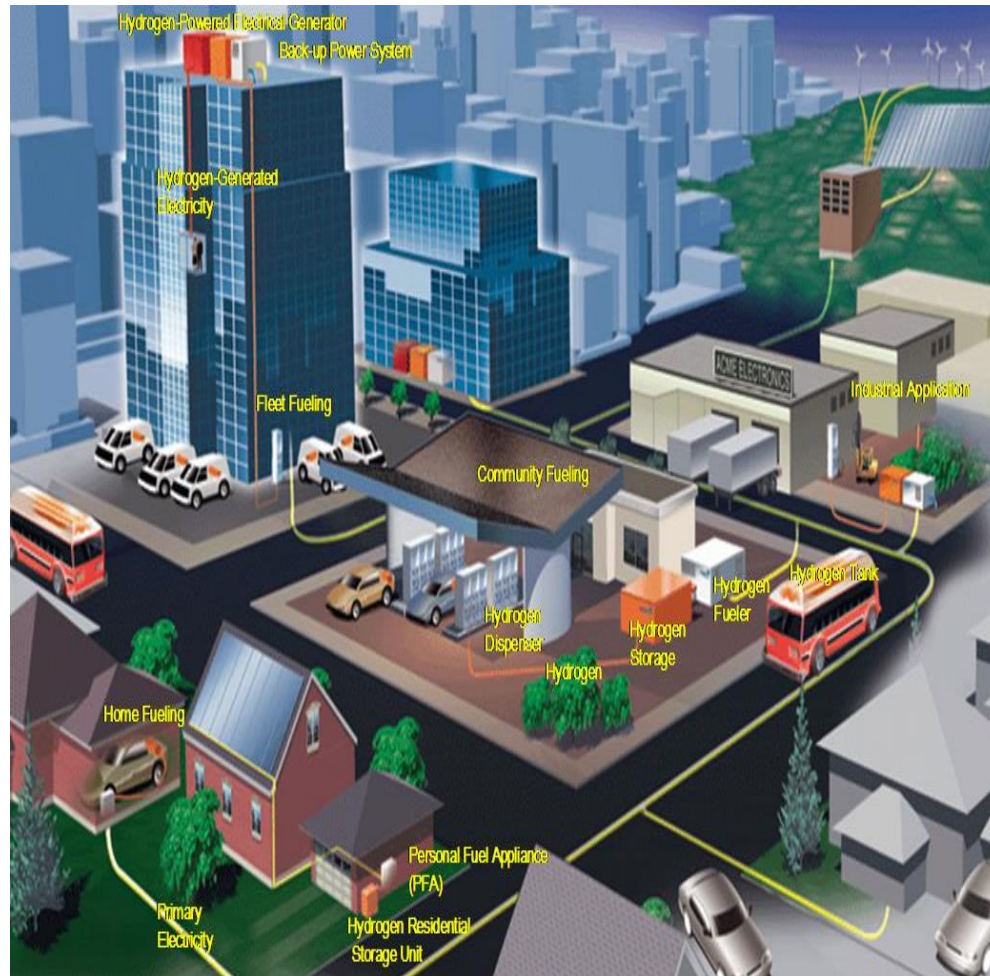


# The Hydrogen Economy – The Future of Energy



**Indian Institute of Technology  
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# Agenda

- A Hydrogen Vision of the Future
- What is Hydrogen Economy?
- Where do we get Hydrogen From?
- Hydrogen Fuel Storing and Transport
- Hydrogen Economy
- Problems with Hydrogen
- The Promise of Hydrogen
- Hydrogen Summary

# A Vision of a Hydrogen Future

*"I believe that water will one day be employed as fuel, that hydrogen and oxygen which constitute it, used singly or together, will furnish an inexhaustible source of heat and light, of an intensity of which coal is not capable. I believe then that when the deposits of coal are exhausted, we shall heat and warm ourselves with water. Water will be the coal of the future."*

Jules Verne (1870) *L'île mystérieuse*

# What Is Hydrogen Economy?

## •Now: “Fossil Fuel Economy”

- Dependencies on foreign oil
- Carbon dioxide = major pollutant & GHG
- Without fossil fuels, economy at a standstill

## •What if we had a “Hydrogen Economy?”

- Clean: Hydrogen fuel cells' only byproduct is water
- Economic independence: can be produced anywhere (even in homes)

# What Is Hydrogen Economy?

The hydrogen economy is a vision (**Hydrogen is clean energy choice Hydrogen is flexible, affordable, safe, domestically produced, used in all sectors of the economy, and in all regions of the country**).

It consists of an economic system in which energy is supplied by renewable resources.

In this “economy” hydrogen is the medium of energy storage and transport.

In other words: Solar, Wind, Hydro, biogas, etc.

National Vision to 2030 and Beyond\*

# *Why Hydrogen ?*

Technical reasons:

- Hydrogen has more energy per mass
- Technology can handle lighter fuels
- Too much carbon in the atmosphere already!

• Other forces:

- market in fuel cells
- visible urban pollution

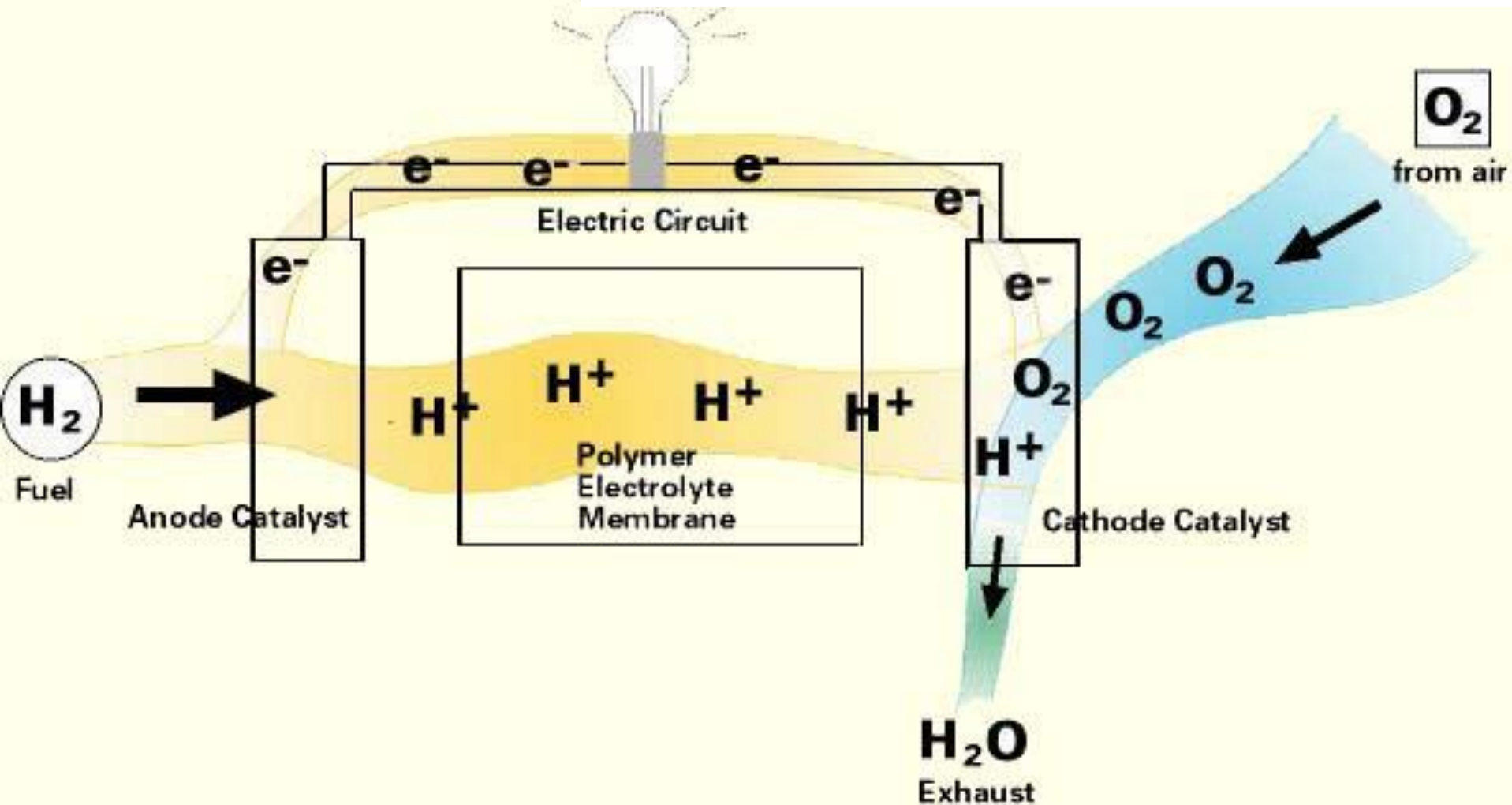
# Where Do You Get Hydrogen?

Hydrogen can be extracted from fossil fuels, including gasoline, diesel, natural gas, propane or coal as needed, or **through electrolysis of water** using electricity from conventional power plants.

- ✓ Hydrogen can be made from renewable resources - from water using electricity from solar. photovoltaic or wind or using direct processes.
- ✓ From methane gas, landfills, waste treatment or biomass processing.

# Fuel Cell

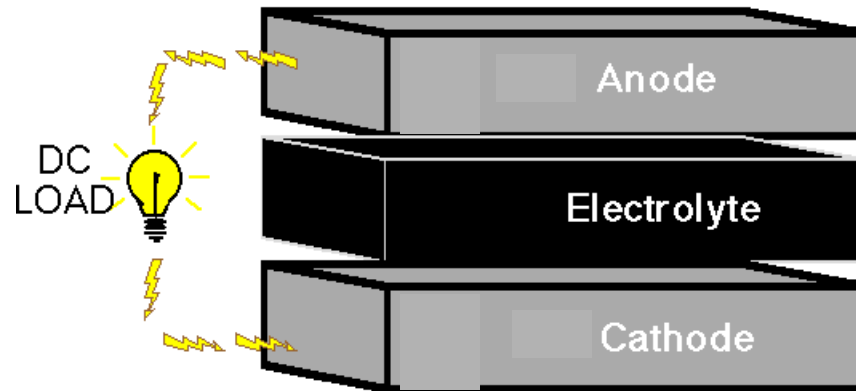
## Where do we get the hydrogen from?





# Fuel Cell

## BASIC FUEL CELL SCHEMATIC



- All fuel cells have the same basic configuration - an electrolyte and two electrodes.
- Fuel cells are classified by the kind of electrolyte used.
- The type of electrolyte used determines the kind of chemical reactions that take place and the temperature range of operation.

## Fuel Cell Type

PEMFC - Polymer Electrolyte Membrane Fuel Cells

(or Proton Exchange Membrane Fuel Cells )

DMFC - Direct Methanol Fuel Cells

AFC - Alkaline Fuel Cells

PAFC - Phosphoric Acid Fuel Cells

MCFC - Molten Carbonate Fuel Cells

SOFC - Solid Oxide Fuel Cells

# Fuel Cell Type

Fuel cell type	Operating Temp (°C)	Efficiency	Suitable applications			
			Domestic power	Small-scale power	Large-scale	Transport
PEMFC	50-120	40-50	√	√	X	√
AFC	50-90	50-70	√	√	X	√
PAFC	150-220	40-45	X	√	X	X
MCFC	600-650	50-60	X	√	√	X
SOFC	800-1000	50-60	√	√	√	X

# Producing Hydrogen

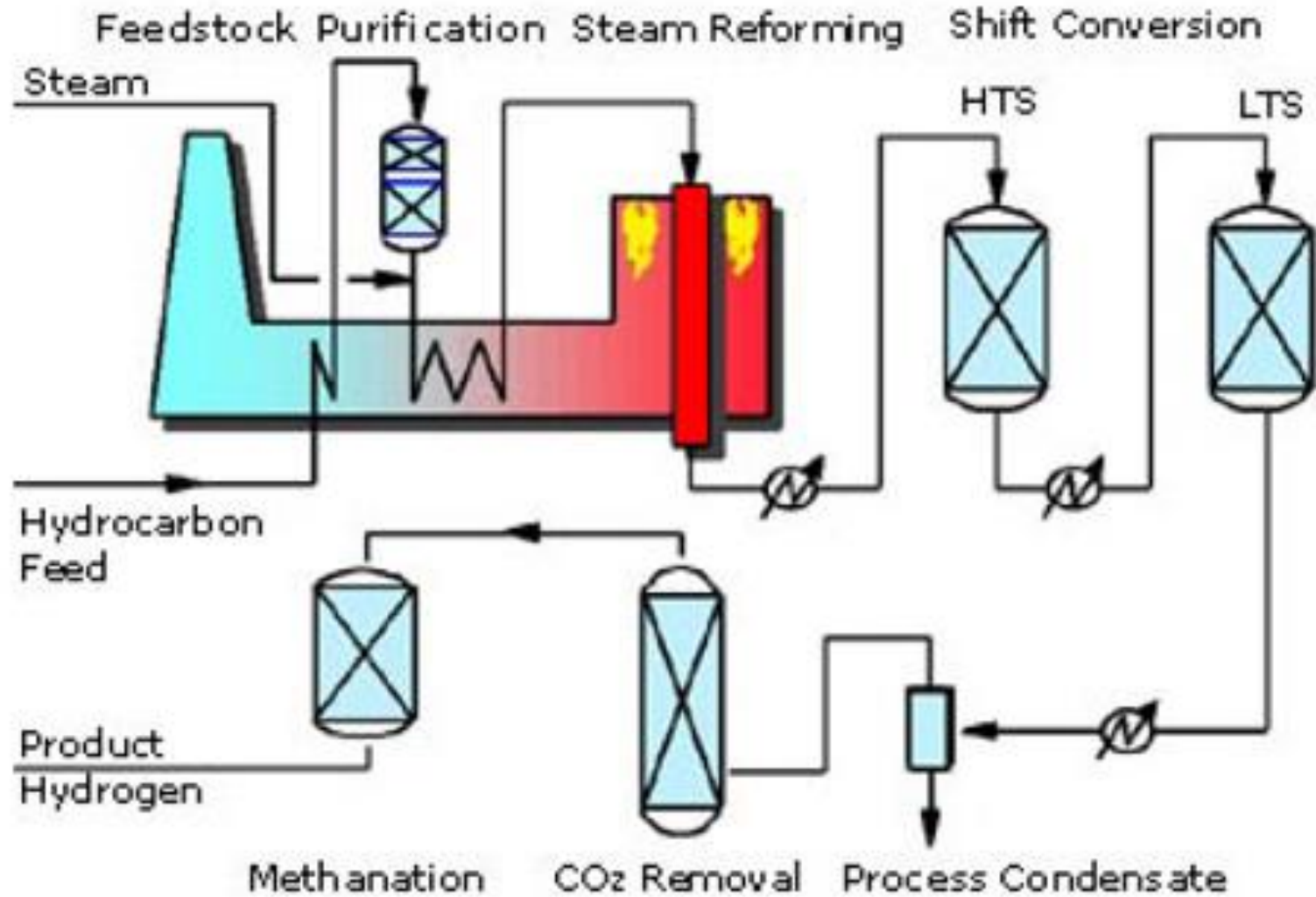
# How is Hydrogen Produced?

- Reforming fossil fuels
  - Heat hydrocarbons with steam
  - Produce  $H_2$  and CO
- Electrolysis of water
  - Use electricity to split water into  $O_2$  and  $H_2$
- High Temperature Electrolysis
  - Experimental
- Biological processes
  - Very common in nature
  - Experimental in laboratories

# Steam Reforming

- From any hydrocarbon
  - Natural gas typically used
- Water (steam) and hydrocarbon mixed at high temperature (700–1100 °C)
  - Steam (H<sub>2</sub>O) reacts with methane (CH<sub>4</sub>)
  - $\text{CH}_4 + \text{H}_2\text{O} \rightarrow \text{CO} + 3 \text{H}_2 - 191.7 \text{ kJ/mol}$
- The thermodynamic efficiency comparable to (or worse than) an internal combustion engine
  - Difficult to motivate investment in technology

## Hydrogen from steam reforming: 95% of the usage



HTS – High temperature shift

LTS – Low temperature shift

# Hydrogen Steam Reforming Plants





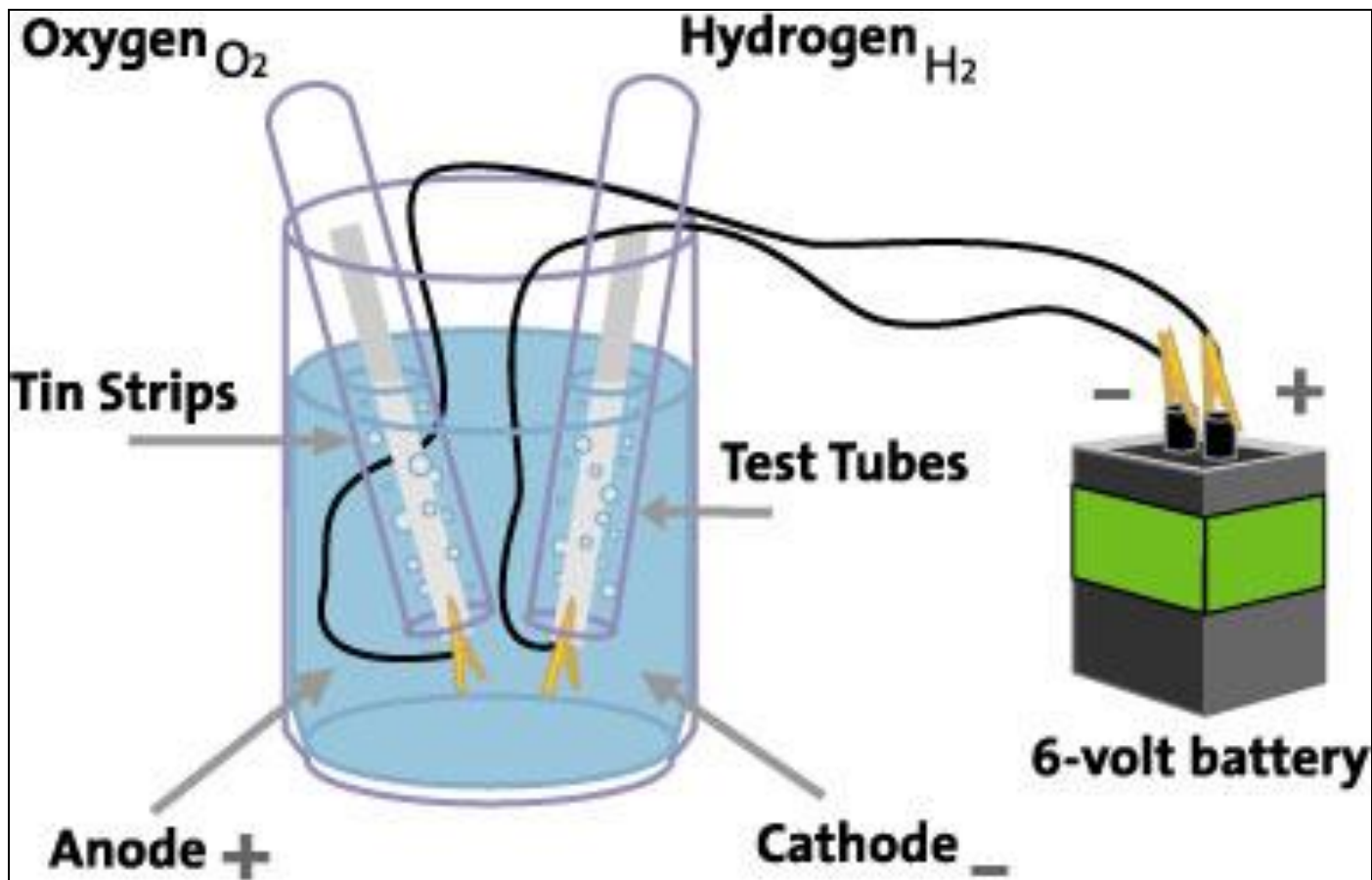
## **Where does the power come from?**

Wind

Solar PV and

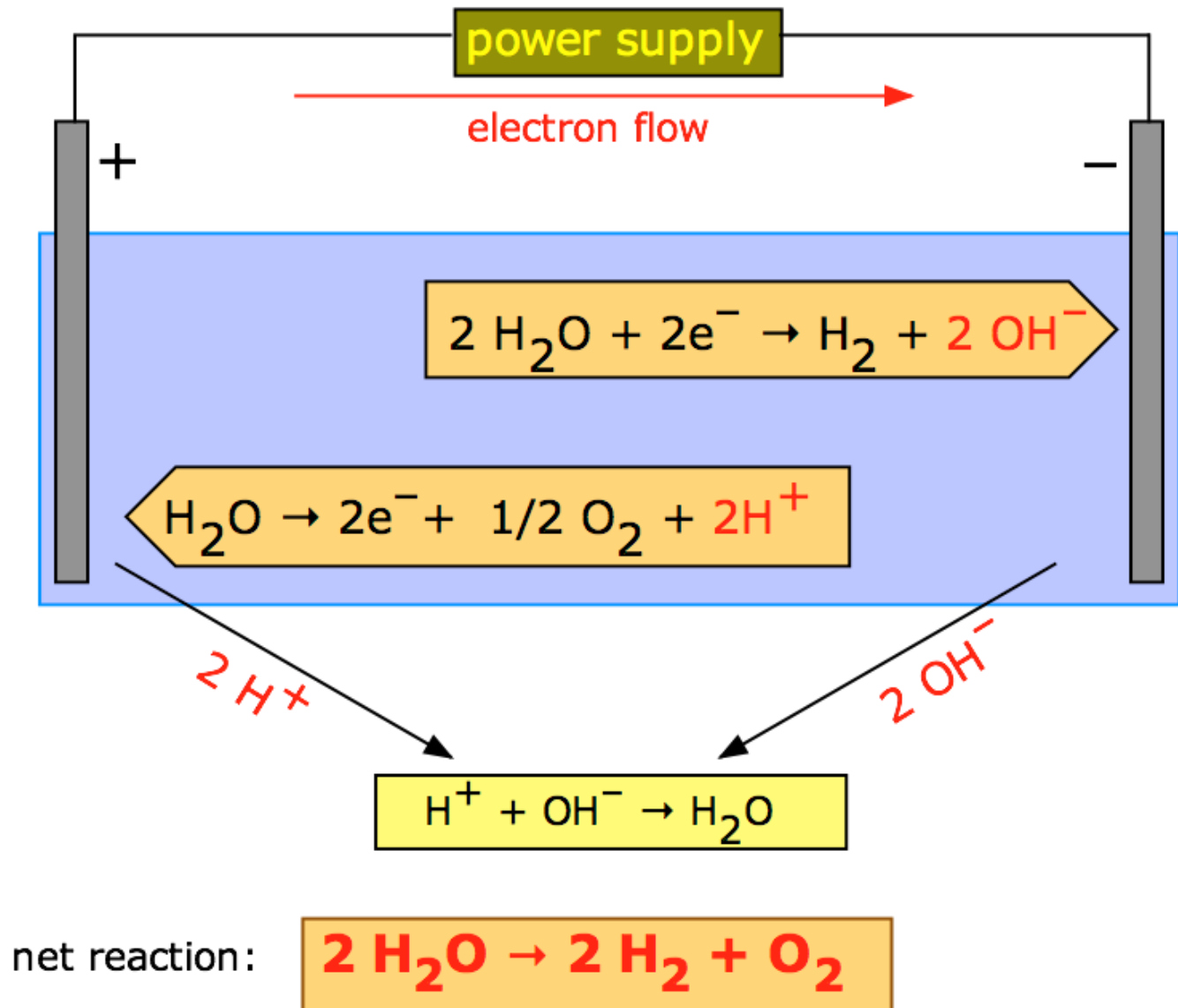
Other renewable Energy Resources.

# Electrolysis of Water ( $\text{H}_2\text{O}$ )

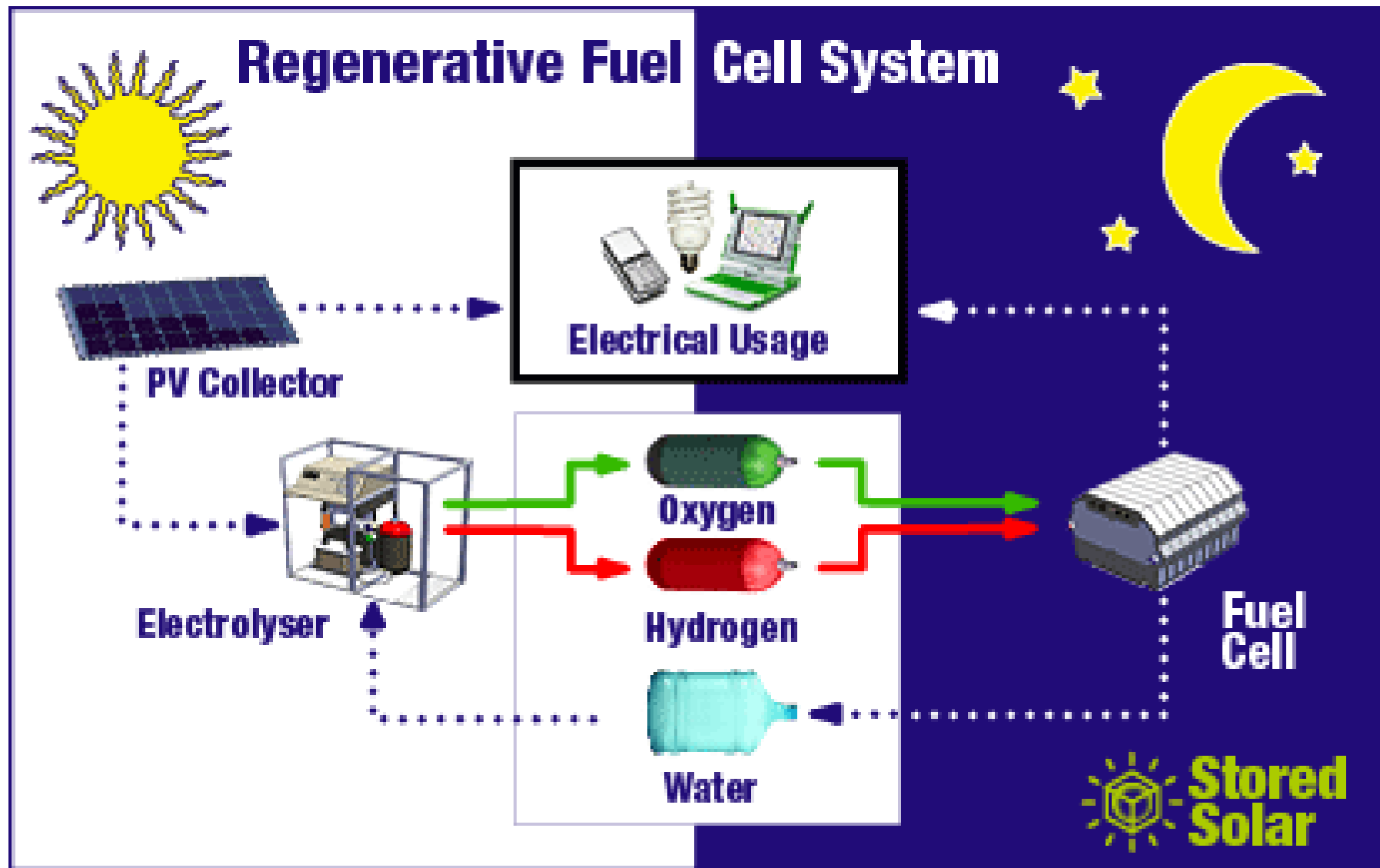


## Hydrogen from electrolysis:

5% of the usage



# Regenerative Fuel Cell



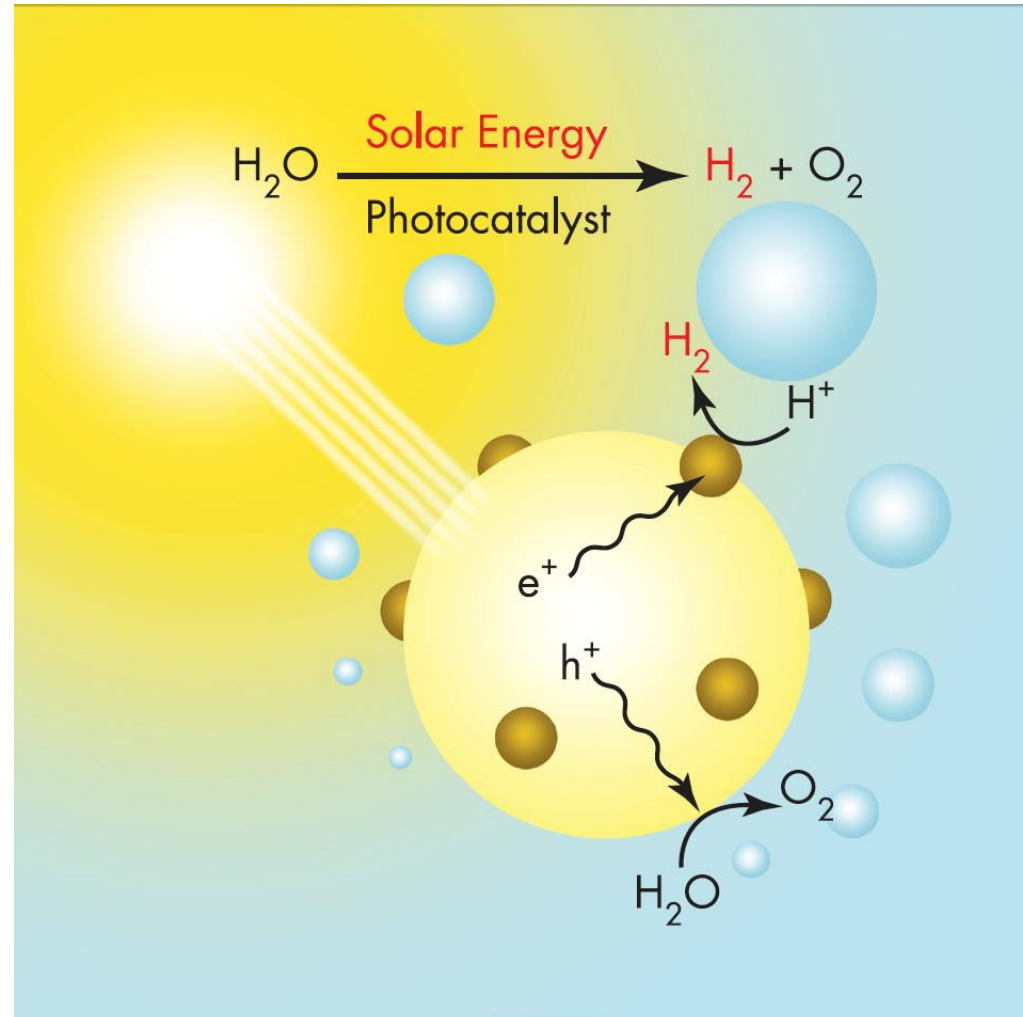
# Renewable Energy for Electrolysis



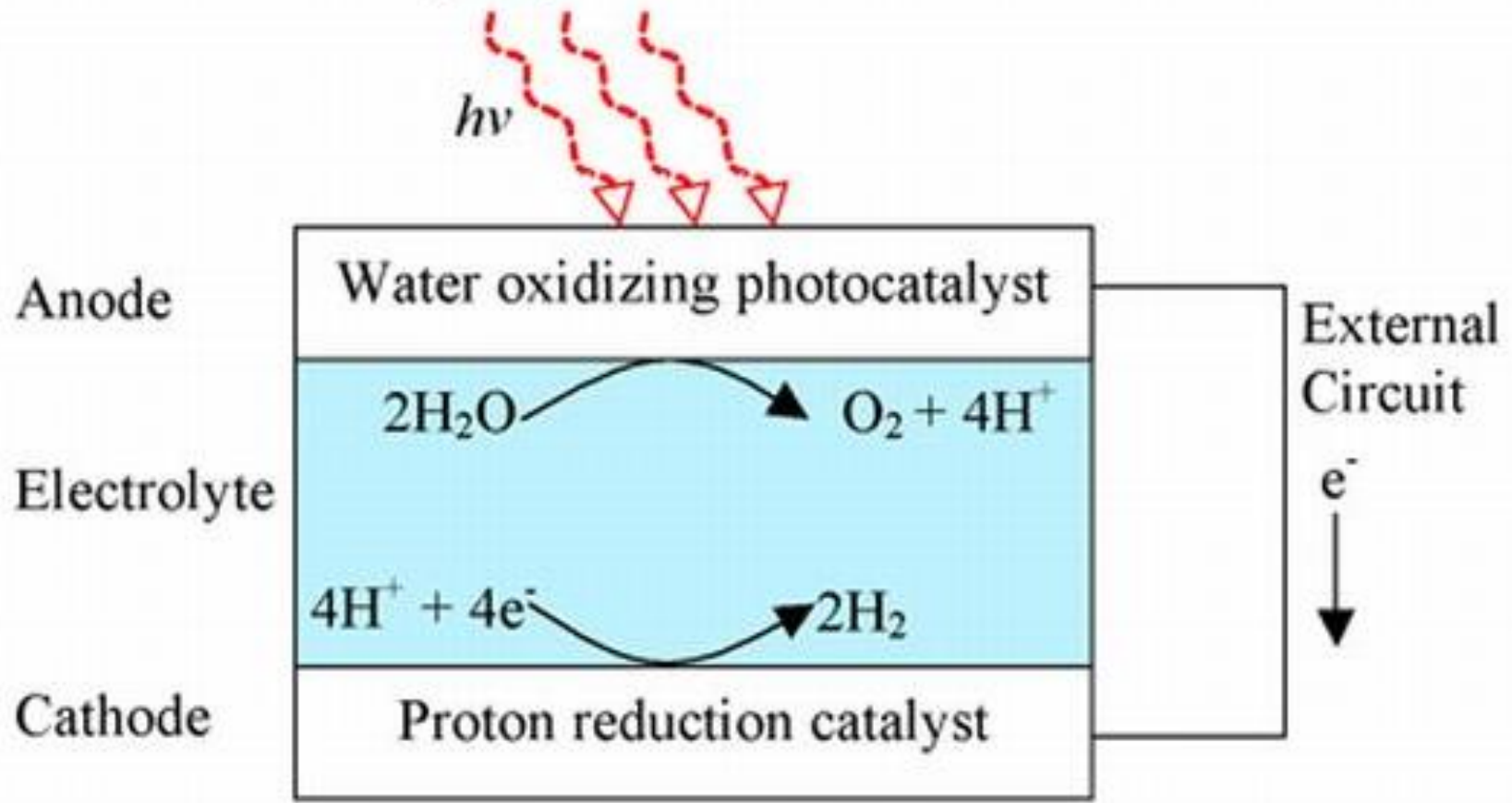
## Hydrogen from water-splitting:

Solar water splitting is the process by which energy in solar photons is used to break down liquid water into molecules of hydrogen and oxygen gas.

Hydrogen produced through solar water does not emit carbon into the atmosphere.



## Schematic of proposed Photoelectrochemical Cell

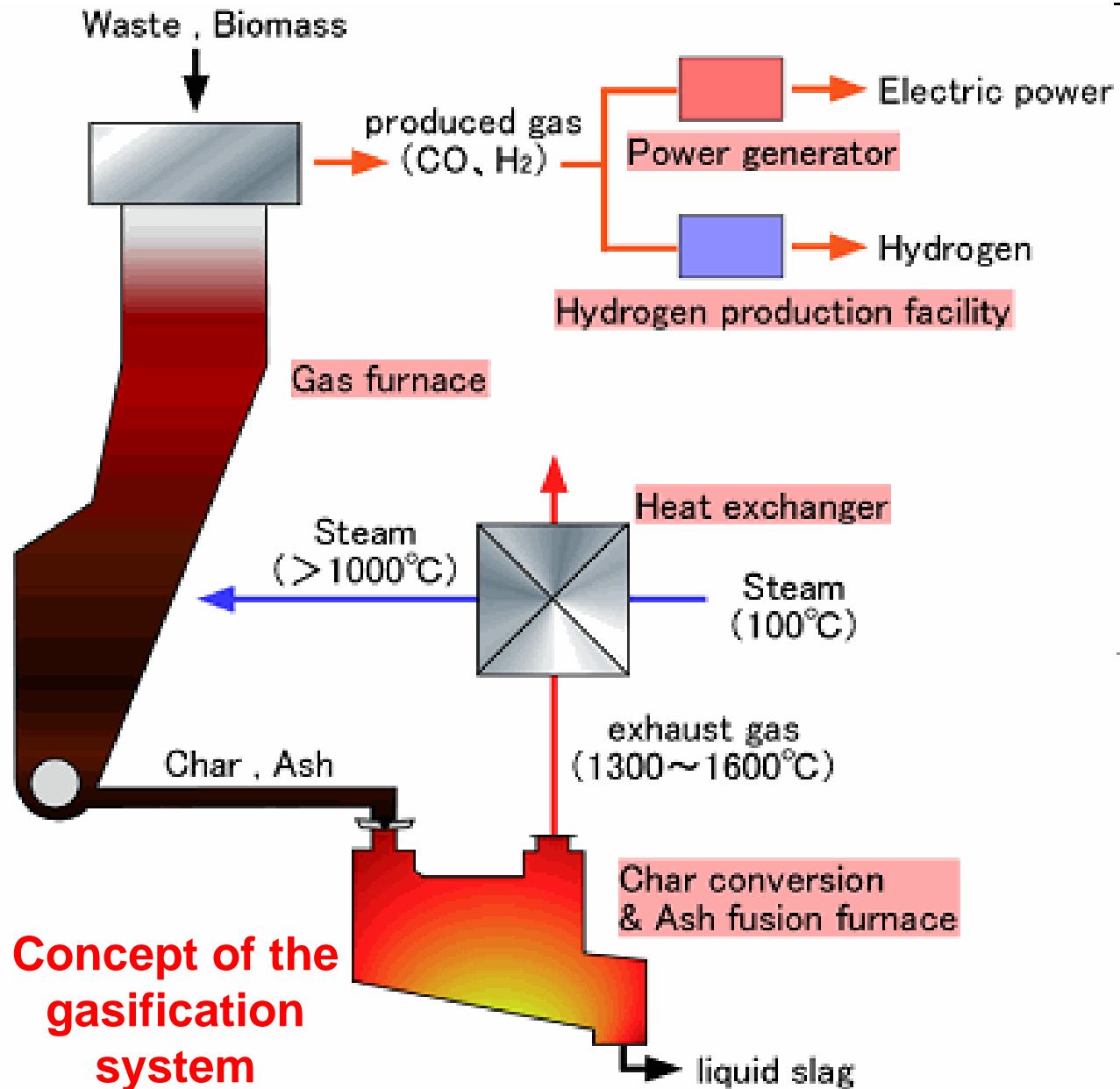


# Fuel Cell

## Hydrogen from waste:



HyPR-MEET  
demonstration plant

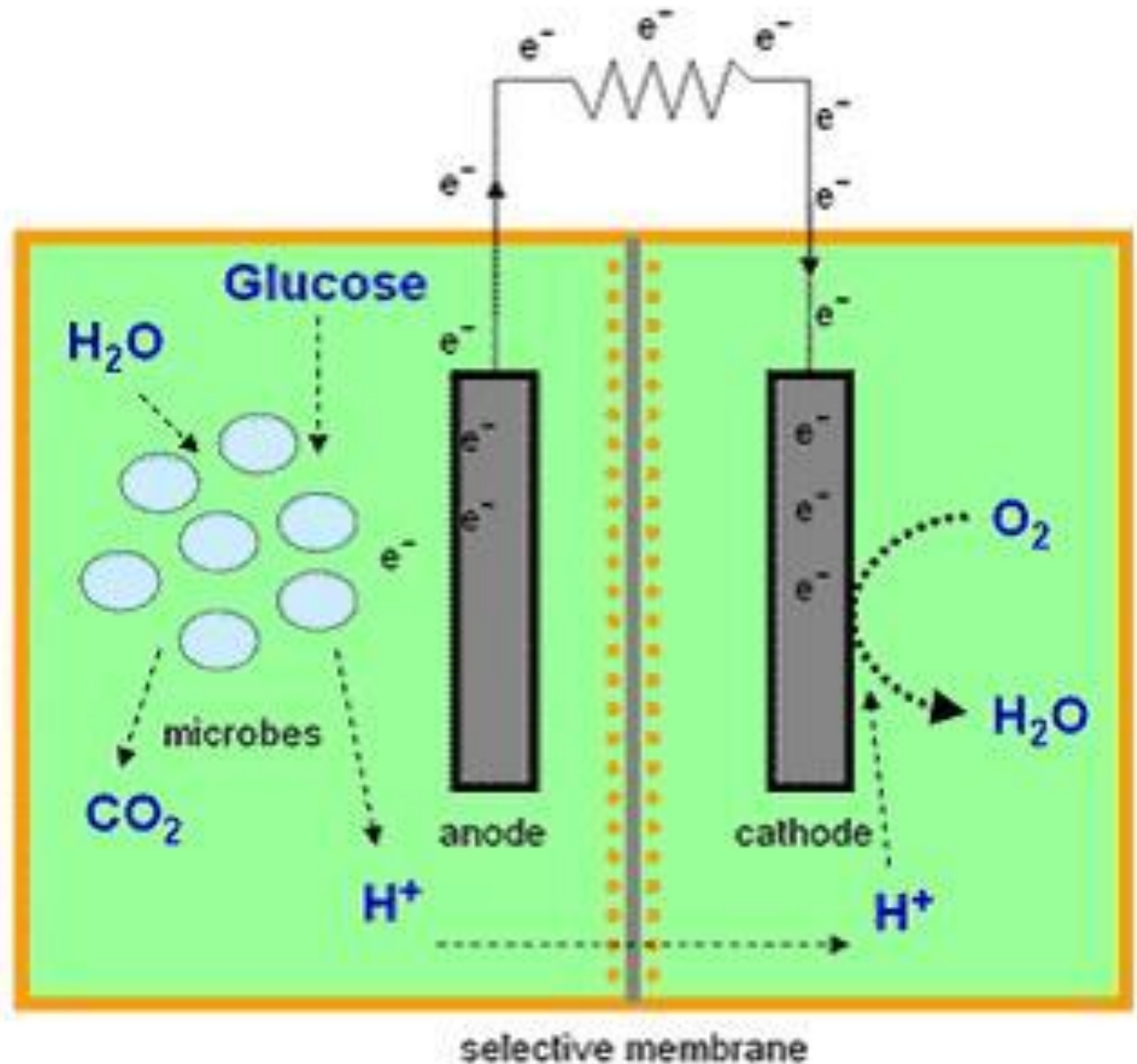




# Biomass Electrolysis Module

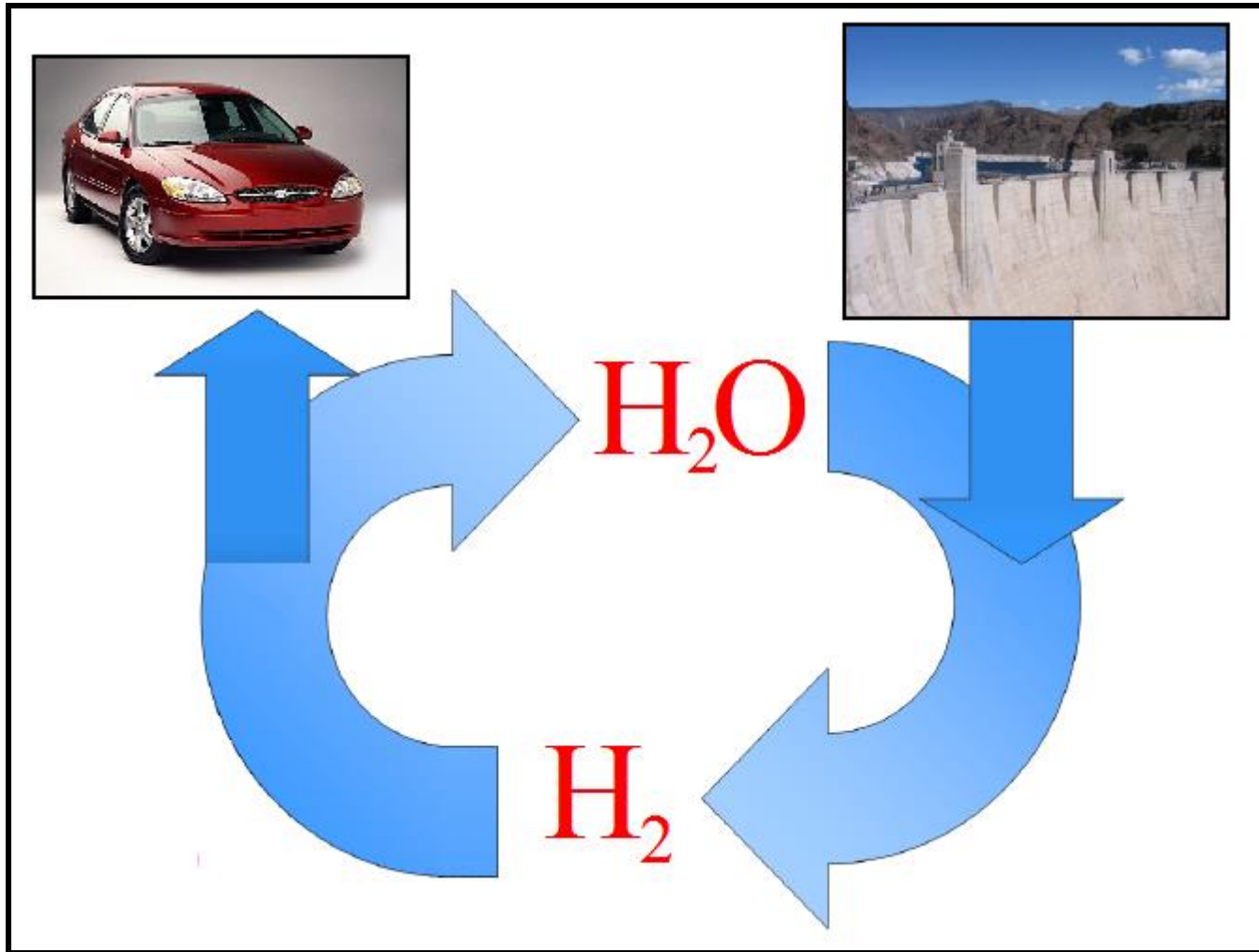


# Microbial Fuel Cells

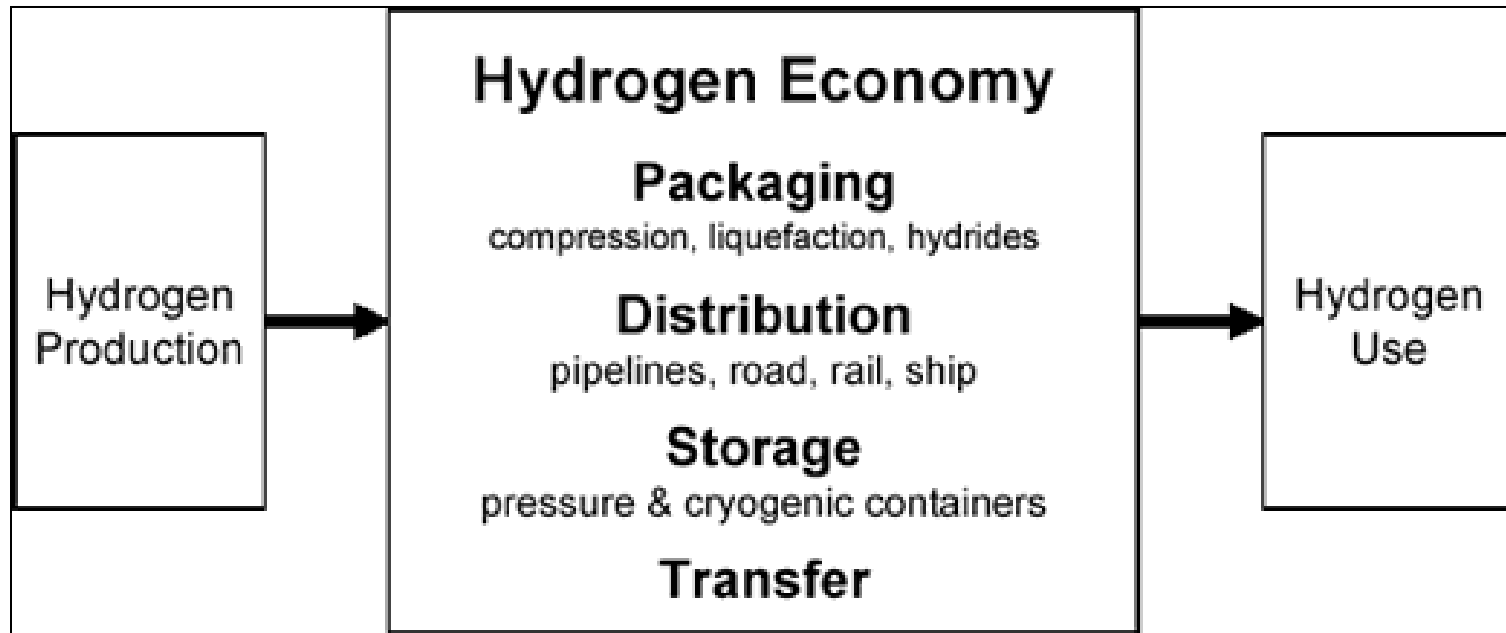


Source: <http://parts.mit.edu/igem07/images/2/2d/Fuelcell.JPG>

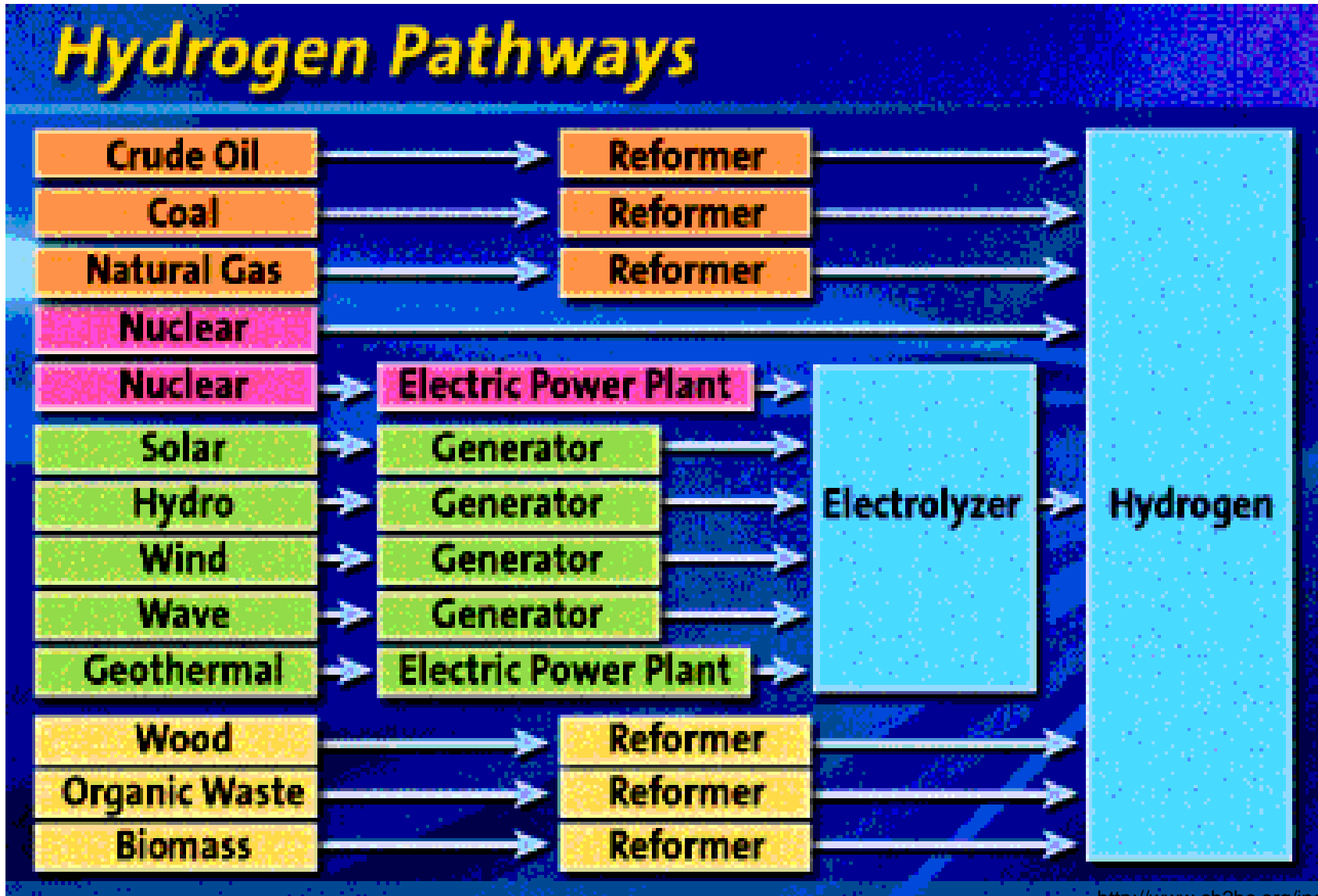
# Hydrogen Energy Cycle



# Operating the Hydrogen Economy



# Hydrogen Pathways



# Advantages of a Hydrogen Economy

- Waste product of burning  $H_2$  is water
- Elimination of fossil fuel pollution
- Elimination of greenhouse gases
- Elimination of economic dependence
- Distributed production

# Issues with Hydrogen

- Not widely available on planet earth
- Usually chemically combined in water or fossil fuels (must be separated)
- Fossil fuel sources contribute to pollution and greenhouse gases
- Electrolysis requires prodigious amounts of energy

# Technological Questions

- Where does hydrogen come from?
- How is it transported?
- How is it distributed?
- How is it stored?



# Direct Methanol Fuel Cell (DMFC)

- Operates at about 50-90°C
- Efficiency is about 40%
- Used more for small portable power applications, possibly cell phones and laptops



Toshiba  
Corporation

# Storing & Transporting Hydrogen

# Storing the Hydrogen

Developing safe, reliable, compact and cost-effective hydrogen storage is one of the biggest challenges to widespread use of fuel cell technology.

# Hydrogen Storage

- Storage a major difficulty with hydrogen
- H<sub>2</sub> has low energy density per volume
  - Requires large tanks to store
- H<sub>2</sub> can be compressed to reduce volume
  - Requires heavy, strong tanks
- H<sub>2</sub> can be liquefied to reduce volume
  - Boils at -423 °F (cryogenic)
  - Requires heavily insulated, expensive tanks
- Both compression and liquefaction require a lot of energy

# Ammonia Storage

- $\text{H}_2$  can be stored as ammonia ( $\text{NH}_3$ )
- Exceptionally high hydrogen densities
- Ammonia very common chemical
  - Large infrastructure already exists
- Easily reformed to produce hydrogen
  - No harmful waste
- BUT
  - Ammonia production is energy intensive
  - Ammonia is a toxic gas

# Metal Hydride Storage

- Metal hydrides can carry hydrogen
  - Boron, lithium, sodium
  - Good energy density, but worse than gas
- Volumes much larger than gasoline
  - Three times more volume
  - Four times heavier
- Hydrides can react violently with water
- Leading contenders
  - Sodium Borohydride
  - Lithium Aluminum Hydride
  - Ammonia Borane

# Transporting Hydrogen



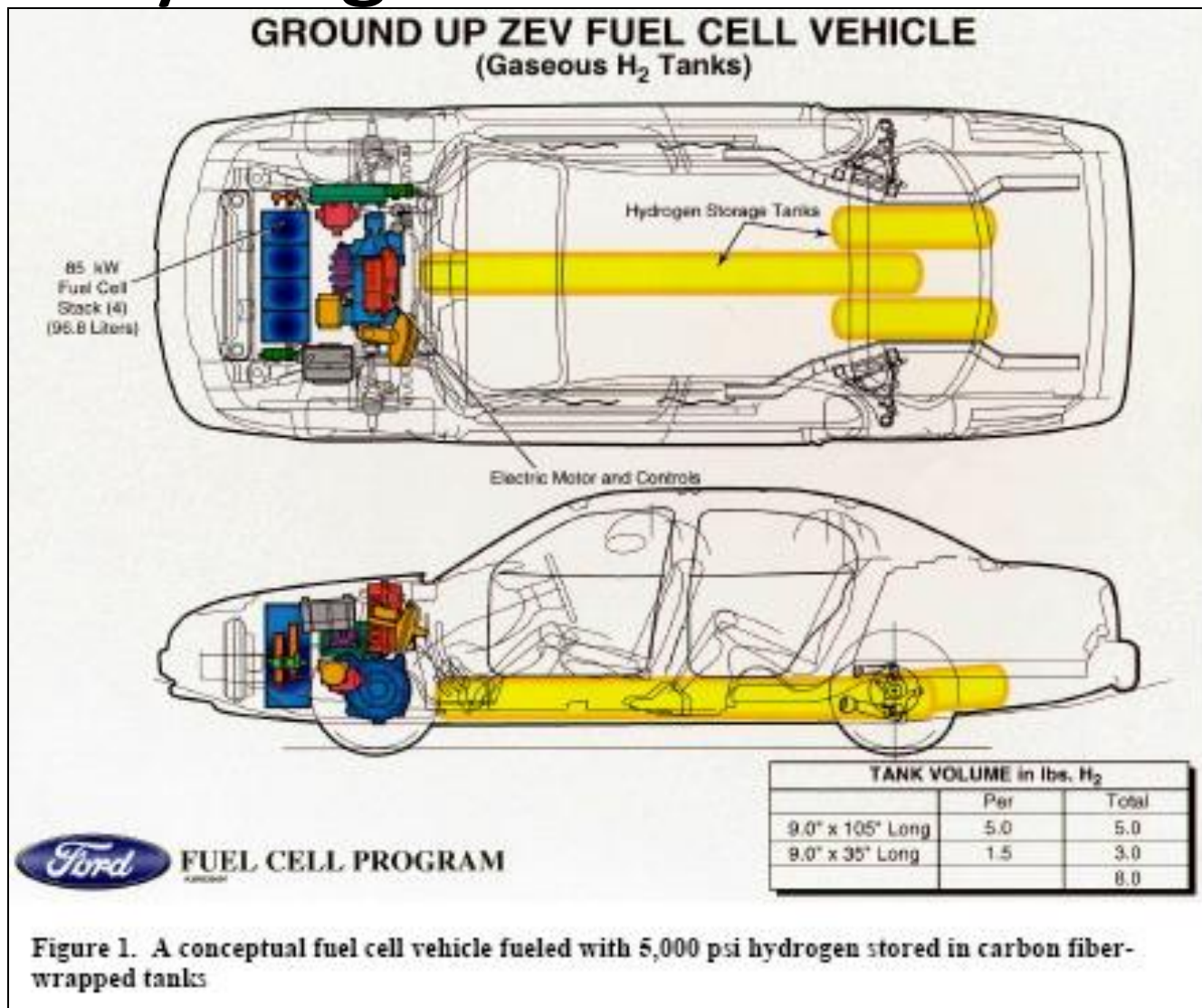
# Storing & Transporting Hydrogen

- Store and Transport as a Gas
  - Bulky gas
  - Compressing  $H_2$  requires energy
  - Compressed  $H_2$  has far less energy than the same volume of gasoline
- Store and Transport as a Solid
  - Sodium Borohydride
  - Calcium Hydride
  - Lithium Hydride
  - Sodium Hydride



# Hydrogen Fueled Transport

# Hydrogen-Powered Autos



# How can Fuel Cell Technology be used?

## Transportation

- All major automakers are working to commercialize a fuel cell car.
- fuel cell buses are currently in use in North and South America, Europe, Asia and Australia
- Trains, planes, boats, scooters, and even bicycles are utilizing fuel cell technology as well



# Hydrogen Fueling Station

**Courtesy: Honda solar hydrogen station (FCX Clarity Car)**





# Hydrogen Fueling Station

Courtesy: Honda solar hydrogen station



# Hydrogen Fueling Station

**Courtesy: Honda Introduces Solar Hydrogen Station on Saitama Prefectural Office Grounds**





# Fuel cell SUV

**Courtesy: Honda Introduces Solar Hydrogen Station on Saitama Prefectural Office Grounds**

*Hyundai ix35 fuel cell SUV with Malmö's famous Turning Torso in the background  
(Source: City of Malmö/Peter Adamsson)*



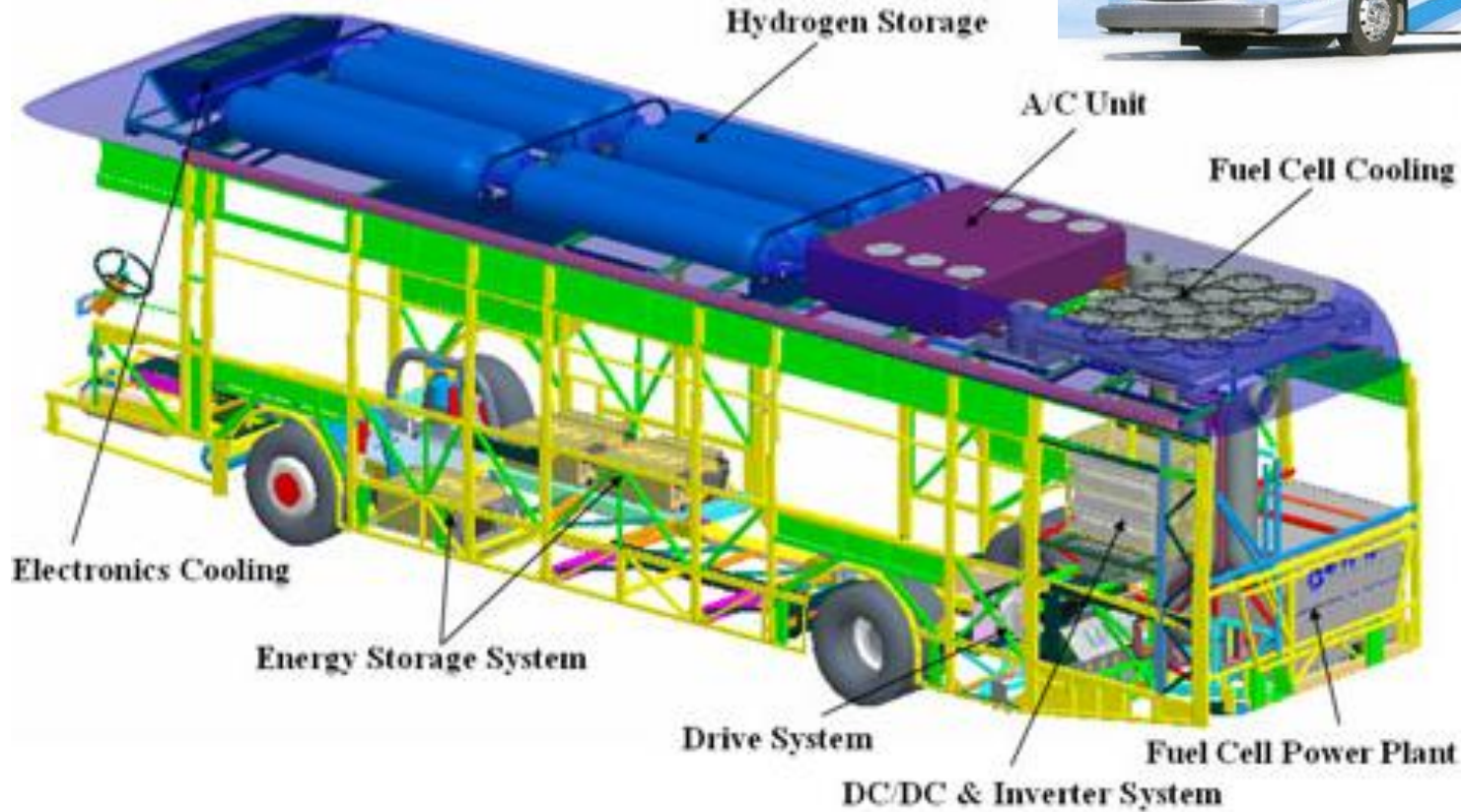
# Hydrogen-Powered Autos





# Fuel Cell Technology For Buses

## Fuel Cell Powered Buses:



28 litres of Hydrogen /100 km  
(compared to 52 litres diesel /100 km)

# Fuel cell BUS

## Hydrogen fuel cell buses in London

### London Open Top Bus Tours



# Hydrogen-Powered Trucks



# Fuel Cell Technology for Planes

## Boeing Flies First Ever Hydrogen Fuel Cell Plane:

The experimental airplane climbed to an altitude of 1,000 m above sea level using a combination of lithium-ion battery power and power generated by hydrogen fuel cells.

After reaching the cruise altitude, batteries were disconnected, and the plane flew straight and level at a cruising speed of 100 km/h for about 20 min on power solely generated by the fuel cells.





# Fuel Cell Technology Aircrafts

## First Commercial Fuel Cell Powered Aircraft:

Airbus and the German Aerospace Center (DLR) presented the first commercial aircraft powered by fuel cells at the ILA Berlin Air Show 2008. The fuel cells cannot replace the plane's jet engines for powering the heavy plane through the air.

Fuel cells replace the auxiliary power units which meet the plane's power demands when the plane is on the ground.



# Hydrogen-Powered Aircraft



Hydrogen powered passenger aircraft with cryogenic tanks along spine of fuselage. Hydrogen fuel requires about 4 times the volume of standard jet fuel (kerosene).

# Fuel Cell Technology For Powered Trains

## Fuel Cell Powered Trains:

Visit <http://hydrail.org/>



# Hydrogen-Powered Rockets





# Nanotechnology in Fuel Cells

- The researchers believe that the improvement is due to a more efficient transfer of electrons than in standard catalysts.
- Increasing catalyst surface area and efficiency by depositing platinum on porous alumina
- Allowing the use of lower purity, and therefore less expensive, hydrogen with an anode made of platinum nanoparticles deposited on titanium oxide.

# Problems with Hydrogen

# Environmental Concerns

- 48% of hydrogen made from natural gas
  - Creates  $\text{CO}_2$  – a greenhouse gas
- Hydrogen  $\text{H}_2$  inevitably leaks from containers
  - Creates free radicals (H) in stratosphere due to ultraviolet radiation
  - Could act as catalysts for ozone depletion

# Hydrogen Economy

The vision of the hydrogen economy is based on two expectations:

(1) that hydrogen can be produced from domestic energy sources in a manner that is affordable and environmentally benign, and

(2) that applications using hydrogen—fuel cell vehicles, for example—can gain market share in competition with the alternatives.

# *“HYDROGEN – The New Source of Power”*

Hydrogen = “Currency of the Future”

HYDROGEN Revolution

Roadmap to the Hydrogen Economy

Why Hydrogen as a Transportation Fuel

- Will Hydrogen be an alternative fuel or the mainstream fuel ?
- Short term and Long term Infrastructure
- Speculation on Timescales

# Hydrogen Economy Schematic

